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Key indicators

Single-crystal X-ray study
T = 220 K
 Mean $\sigma(\text{C}-\text{C}) = 0.008 \text{ \AA}$
 Disorder in main residue
R factor = 0.030
wR factor = 0.077
 Data-to-parameter ratio = 16.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

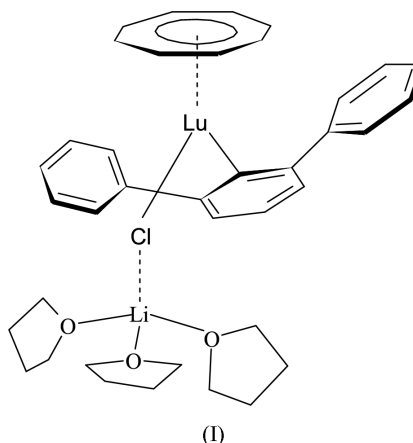
$(\eta^8\text{-Cyclooctatetraenyl})(m\text{-terphenyl})\text{lutetium-}\mu\text{-chloro-}[\text{tris}(\text{tetrahydrofuran})\text{lithium(I)}]$

The molecular structure of a mixed terphenyl COT compound (COT is cyclooctatetraene) of the element lutetium, of composition $\text{DppLuCOT}(\mu\text{-Cl})\text{Li}(\text{THF})_3$ (Dpp is 2,6-diphenylphenyl, or *m*-terphenyl, and THF is tetrahydrofuran) or $[\text{LuLi}(\text{C}_8\text{H}_8)(\text{C}_{18}\text{H}_{13})\text{Cl}(\text{C}_4\text{H}_8\text{O})_3]$, is reported. The monomeric complex contains one η^8 -bonded COT and one σ -bonded terphenyl ligand, as well as a lithium chloride moiety, with the alkali metal atom solvated by three tetrahydrofuran molecules.

Comment

We were interested in investigating the accessibility of mixed terphenyl COT compounds (COT is cyclooctatetraene) of the lanthanides. Our previous studies have shown that such compounds are, in principle, accessible for the lanthanide element samarium (Rabe *et al.*, 2003) in a one-pot synthesis.

The average $\text{Lu}-\text{C}_{\text{COT}}$ distance of (I) is 2.513 Å, while the Lu -centroid distance is 1.74 (4) Å, with centroid-Lu-Cl and centroid-Lu-C9 angles of 128.7 (4) and 133.5 (4)°, respectively. In addition to the $\text{Lu}-\text{C}_9$ distance of 2.394 (4) Å, the next closest Lu -terphenyl distances are to C20 [3.243 (4) Å] and C22 [3.270 (4) Å], indicating possible weak allyl-like interactions. The dihedral angles between the central phenyl ring (C9-C14) and outer (C15-C20 and C21-C26) benzene rings are 38.2 (2) and 46.6 (2)°, respectively. The molecular structure of (I) is isostructural with its samarium analog (Rabe *et al.*, 2003).



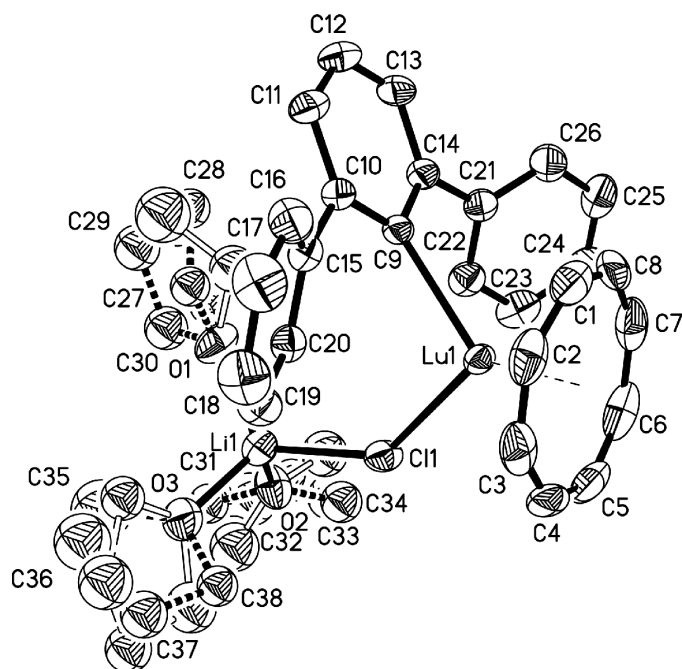
Experimental

Single crystals of (I) were obtained from the one-pot reaction of equimolar amounts of K_2COT (Katz, 1960), LuCl_3 and DppLi (Rabe *et al.*, 2000) in tetrahydrofuran at room temperature, followed by

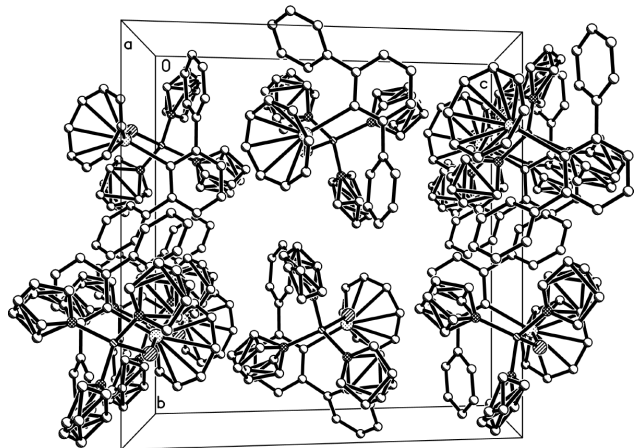
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Figure 1

The molecular structure of the title compound, showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms have been omitted for clarity. C atoms of the tetrahydrofuran groups were refined isotropically. Heavy dashed and open lines are used to highlight the disorder with respect to C atoms of the tetrahydrofuran groups. For clarity, only one set of C atoms of each tetrahydrofuran ring is labeled.


Figure 2

Unit-cell contents, viewed along the *a* axis. All H atoms have been omitted for clarity.

extraction of the crude product with toluene in the presence of a few drops of tetrahydrofuran and cooling to 248 K.

Crystal data

[LuLi(C₈H₈)(C₁₈H₁₃)Cl(C₄H₈O)₃]

M_r = 767.10

Monoclinic, *P*2₁/*n*

a = 12.6936 (19) Å

b = 17.023 (3) Å

c = 16.809 (3) Å

β = 102.843 (3)°

V = 3541.2 (10) Å³

Z = 4

D_x = 1.439 Mg m⁻³

Mo *K*α radiation

Cell parameters from 500

reflections

θ = 2.5–25.0°

μ = 2.90 mm⁻¹

T = 220 (2) K

Block, colorless

0.25 × 0.20 × 0.20 mm

Data collection

Bruker SMART CCD area-detector diffractometer

φ and ω scans

Absorption correction: multi-scan (SADABS; Bruker, 2001)

T_{min} = 0.499, *T_{max}* = 0.559

20 849 measured reflections

6220 independent reflections

4622 reflections with *I* > 2σ(*I*)

R_{int} = 0.032

θ_{max} = 25.0°

h = −15 → 14

k = −19 → 20

l = −19 → 19

Refinement

Refinement on *F*²

R [*F*² > 2σ(*F*²)] = 0.030

wR (*F*²) = 0.077

S = 1.02

6220 reflections

388 parameters

H-atom parameters constrained

w = 1/[σ²(*F_o*²) + (0.0252*P*)² + 3.3351*P*]

where *P* = (*F_o*² + 2*F_c*²)/3

(Δ/σ)_{max} = 0.001

Δρ_{max} = 0.48 e Å⁻³

Δρ_{min} = −0.53 e Å⁻³

Table 1

Selected geometric parameters (Å, °).

Lu1—C9	2.394 (4)	Lu1—C1	2.515 (5)
Lu1—C7	2.482 (5)	Lu1—C4	2.560 (5)
Lu1—C2	2.481 (5)	Lu1—C5	2.562 (5)
Lu1—C3	2.486 (5)	Lu1—Cl1	2.6158 (11)
Lu1—C6	2.505 (5)	Cl1—Li1	2.336 (8)
Lu1—C8	2.510 (5)		
C9—Lu1—C1	97.77 (9)	C14—C9—Lu1	120.0 (3)
Li1—Cl1—Lu1	138.4 (2)	C10—C9—Lu1	122.5 (3)

Atoms C27–C30, C31–C34 and C35–C38 of the tetrahydrofuran rings were found to be disordered over two positions with percentage occupancies of 58 (1)/42 (1), 50 (1)/50 (1) and 28.1 (9)/71.9 (9), respectively. All the C–C bond lengths of the tetrahydrofuran groups were restrained to 1.48 (2) Å, and the O3–C35, O3–C35A, O3–C38 and O3–C38A bond lengths were restrained to 1.40 (2) Å. The distance between C27A and C30A was restrained to 2.20 (2) Å to avoid bumping of these atoms. Even though the disorder was modeled with isotropic displacement parameters for the tetrahydrofuran C atoms, atoms C28A, C32 and C35 of the tetrahydrofuran rings still showed large displacement parameters. All H atoms were placed in calculated positions (0.94–0.98 Å), with isotropic displacement parameters fixed at 1.2 or 1.5 times *U_{eq}* of the parent atom, and were refined as riding atoms.

Data collection: SMART (Bruker, 2001); cell refinement: SMART; data reduction: SAINT (Bruker, 2001) and SHELXTL (Bruker, 2001); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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